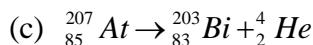
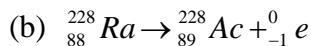
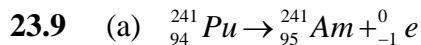
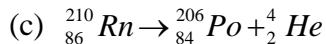
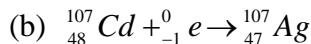
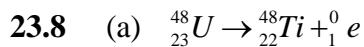
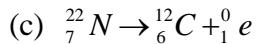
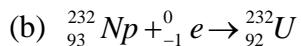
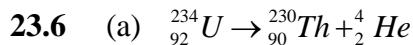


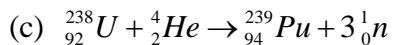
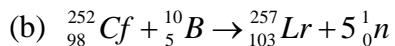
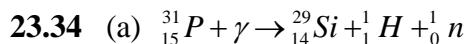
**CHEM 1423**  
**Chapters 23**  
**Homework Answers**

**TEXTBOOK HOMEWORK**



- 23.12** (a)  $\alpha$  decay  
(b) Either positron decay or electron capture  
(c) Either positron decay or electron capture

- 23.13** (a)  $\beta$  decay  
(b)  $\beta$  decay  
(c)  $\beta$  decay



## SUPPLEMENTARY HOMEWORK

- S1.** (a)  $^{31}\text{P}$ :  $E_b = 2.61 \times 10^{10} \text{ kJ/mol}$ ,  $E_b/N = 8.42 \times 10^8 \text{ kJ/mol}$   
(b)  $^{190}\text{Os}$ :  $E_b = 1.51 \times 10^{11} \text{ kJ/mol}$ ,  $E_b/N = 7.96 \times 10^8 \text{ kJ/mol}$   
(c)  $^{239}\text{Pu}$ :  $E_b = 1.81 \times 10^{11} \text{ kJ/mol}$ ,  $E_b/N = 7.57 \times 10^8 \text{ kJ/mol}$
- S2.** (a)  $\Delta E = -1.1 \times 10^{10} \text{ kJ/mol}$   
(b)  $\Delta E = -1.6 \times 10^9 \text{ kJ/mol}$   
(c)  $\Delta E = -1.8 \times 10^8 \text{ kJ/mol}$